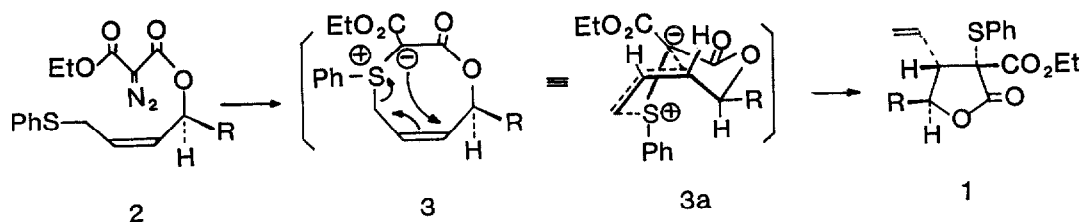


STEREOSELECTIVE SYNTHESIS OF CONTIGUOUSLY SUBSTITUTED BUTYRO-
LACTONES BASED ON THE CYCLIC ALLYLSULFONIUM YLIDE REARRANGEMENT

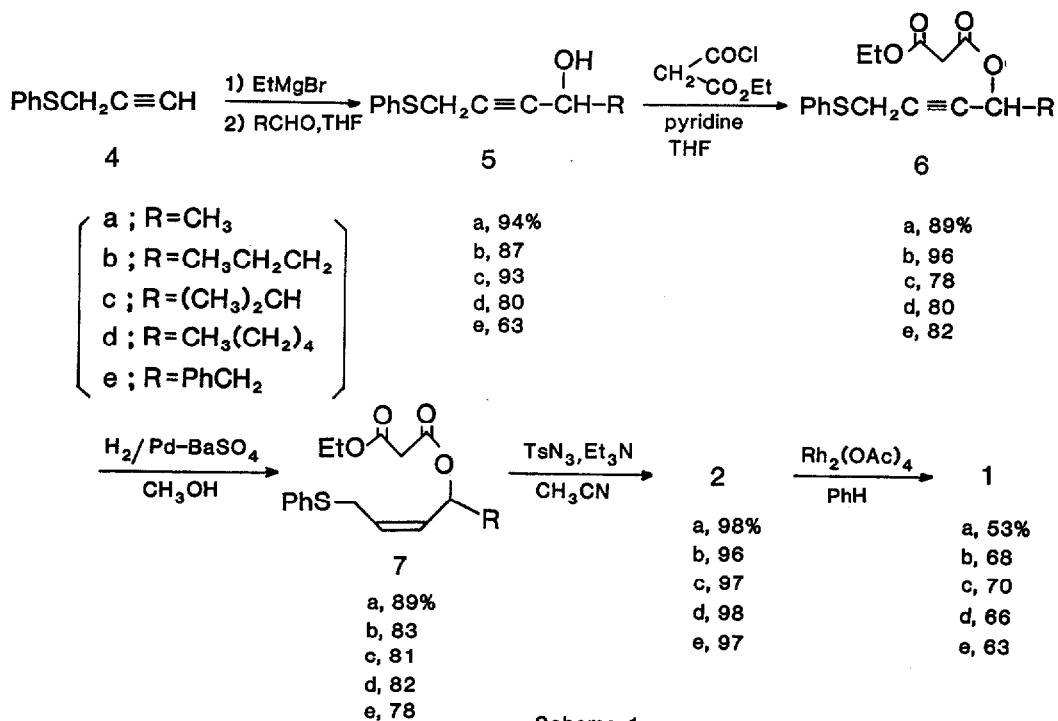
Fusao Kido, Subhash C. Sinha, Toshiya Abiko, and Akira Yoshikoshi*
Chemical Research Institute of Non-Aqueous Solutions,
Tohoku University, Sendai 980, Japan.

Summary: α -Diazomalonates of (*Z*)-4-phenylthio-2-buten-1-ol homologues (**2**) stereoselectively provided γ -alkyl- α -ethoxycarbonyl- α -phenylthio- β -vinylbutyrolactones (**1**) by the [2.3]sigmatropic rearrangement of cyclic sulfonium ylides (**3**) intramolecularly generated therefrom with rhodium acetate.

The [2.3]sigmatropic rearrangement of allylic sulfonium ylides has recently been received much attention in organic chemistry.¹ In this reaction, rhodium acetate has been found to be a catalyst that is particularly efficient for the generation of sulfonium ylides from sulfides and diazo compounds.² In connection with an ongoing project in this laboratory, we wish to report here the highly stereoselective synthesis of contiguously substituted γ -butyrolactones (**1**) by the [2.3]sigmatropic rearrangement of eight-membered cyclic sulfonium ylides (**3**) which were generated by intramolecular addition from diazo sulfides (**2**) in the presence of rhodium acetate.



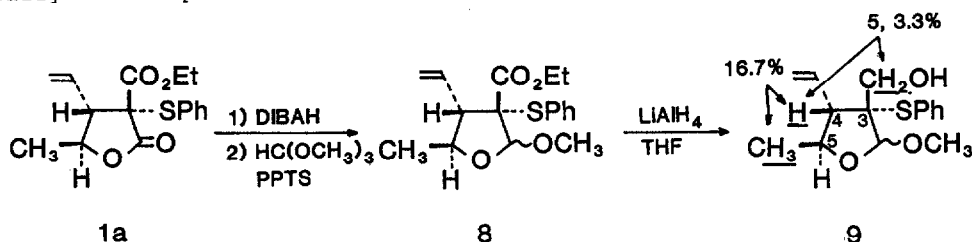
Z-Olefins (**2**) containing both diazo and sulfide groups were prepared from propargyl bromide by a 6-step synthetic sequence as follows (Scheme 1); treatment of phenyl propargyl sulfide (**4**),³ readily obtainable from propargyl bromide in benzene and aqueous sodium thiophenoxide in the presence of a phase transfer catalyst (tetrabutylammonium hydroxide) in quantitative yield,⁴ with ethylmagnesium bromide in tetrahydrofuran followed by addition of aldehydes at 0 °C provided secondary alcohols **5a-5e** in excellent yields. Esterification of these alcohols with ethyl malonyl chloride and pyridine in tetrahydrofuran at 0 °C for 3 h afforded the malonates **6a-**



6e in high yields.⁵ Selective hydrogenation of the triple bonds in these malonates on 5% Pd-BaSO₄ provided the desired Z-olefins **7a-7e** without formation of the corresponding E-isomers. Reaction of **7a-7e** with p-toluenesulfonyl azide and triethylamine in acetonitrile at 45 °C for 40 h yielded diazo esters **2a-2e** in excellent yields.⁵ While several attempts for the generation of carbenes from these diazo esters by employing a typical copper catalyst such as copper powder or cupric sulfate turned out the recovery of substrates or formation of intractable mixtures, treatment of the diazo esters **2a-2e** with a catalytic amount of rhodium acetate (0.01 equiv.),⁶ freshly prepared from rhodium chloride,⁷ in refluxing benzene, afforded γ-butyrolactones **1a-1e**⁸ as single stereoisomers in moderate to good yields.

The stereochemistry of substituents on the lactone ring in **1a-1e** was assigned based on NOE experiments of compound **9**, which was derived from **1a** by diisobutylaluminum hydride reduction, acetalization of the resulting hemiacetal, followed by lithium aluminum hydride reduction (Scheme 2). A large NOE was observed between C(4)-H (δ 3.17) and C(5)-CH₃ (δ 1.33) and between C(4)-H and methylene protons (δ 3.74 and 3.87) of C(3)-CH₂OH in **9**. These results obviously demonstrate that an alkyl group (R) orients equa-

torially in the preferred transition state (3a).



Scheme 2

The experimental procedure for the rearrangement is as follows; after a solution of **2** (1 mmol) and freshly prepared rhodium acetate⁷ in benzene (6--8 mL) had been stirred for 10 min at room temperature, it was heated for 20 min at 80 °C. The resulting mixture was passed through a short silica gel column and removal of the solvent afforded crude **1**, which was purified by HPLC.

To our knowledge, the [2.3]sigmatropic rearrangement via eight-membered cyclic sulfonium ylide seems to be unprecedented. The products obtained here would be useful precursors for the corresponding 2,5-dihydro-5-alkyl-4-vinyl-2-furanone derivatives whose analogues have successfully been employed as building blocks in natural product synthesis.⁹

Acknowledgment. We are indebted to Prof. K. Ogasawara (Pharmaceutical Institute of this University) for 500 MHz NMR measurements. This work was supported by a Grant-in-Aid for Special Project Research (Chemical Syntheses for Elucidation of Biological Functions).

References and Notes

1. Trost, B. M.; Melvin, Jr. L. S. "Sulfur Ylides", Academic Press, New York, 1975, pp 108-127 and references.
2. Doyle, M. P. *Acc. Chem. Res.*, 1986, **19**, 348.
3. All new compounds reported herein exhibited reasonable spectral properties and gave satisfactory microanalytical data.
4. (Alternative preparation) Bakuzis, P.; Bakuzis, M. L. F.; Fortes, C. C.; Santos, R. *J. Org. Chem.*, 1976, **41**, 2769.
5. Corey, E. J.; Fuchs, P. L. *J. Am. Chem. Soc.*, 1972, **94**, 4014.
6. Takano, S.; Tomita, S.; Takahashi, M.; Ogasawara, K. *Chem. Lett.*, 1987, 1569.
7. Legzdins, P.; Mitchell, R. M.; Rempel, G. L.; Ruddick, J. D.; Wilkinson, G. *J. Chem. Soc., A*, 1970, 3322.

8. (^1H NMR data at 90 MHz) **1a** 7.20-7.70 (m, 5H), 5.95 (ddd, 1H, \underline{J} =18, 10.8, 8), 5.42 (dd, 1H, \underline{J} =10.8, 1), 5.34 (dd, 1H, \underline{J} =18, 1), 4.73 (dq, 1H, \underline{J} =10, 6.5), 4.28 (q, 2H, \underline{J} =6.5), 3.36 (dd, 1H, \underline{J} =10, 8), 1.46 (d, 3H, \underline{J} =6.5), 1.29 (t, 3H, \underline{J} =6.5); **1b** 7.20-7.74 (m, 5H), 5.96 (ddd, 1H, \underline{J} =17, 10.8, 8), 5.39 (d, 1H, \underline{J} =10.8), 5.30 (dd, 1H, \underline{J} =17, 2), 4.57 (ddd, 1H, \underline{J} =10, 7.2, 3.6), 4.26 (q, 2H, \underline{J} =6.5), 3.40 (dd, 1H, \underline{J} =10.8, 8), 1.40-1.72 (m, 4H), 1.26 (t, 3H, \underline{J} =6.5), 0.96 (t, 3H, \underline{J} =5); **1c** 7.20-7.68 (m, 5H), 5.96 (ddd, 1H, \underline{J} =18, 10.8, 8), 5.37 (d, 1H, \underline{J} =10.8), 5.30 (dd, \underline{J} =18, 2), 4.46 (dd, 1H, \underline{J} =10, 5), 4.21 (q, 2H, \underline{J} =6.5), 3.56 (dd, 1H, \underline{J} =10, 8), 1.94 (m, 1H), 1.26 (t, 3H, \underline{J} =6.5), 1.04, 1.01 (d, 3H, \underline{J} =6 each); **1d** 7.20-7.72 (m, 5H), 5.96 (ddd, 1H, \underline{J} =18, 10.8, 8), 5.39 (d, 1H, \underline{J} =10.8), 5.30 (dd, \underline{J} =10.8, 2), 4.56 (m, H), 4.24 (q, 2H, \underline{J} =6.5), 3.40 (dd, 1H, \underline{J} =10, 8), 1.24-1.76 (m, 8H), 1.25 (t, 3H, \underline{J} =6.5), 0.90 (t, 3H, \underline{J} =6.5); **1e** 7.20-7.68 (m, 10H), 5.97 (ddd, \underline{J} =18, 10.8, 8), 5.46 (d, 1H, \underline{J} =10.8), 5.30 (dd, 1H, \underline{J} =18, 2), 4.82 (ddd, 1H, \underline{J} =10, 7, 4), 4.20 (q, 2H, \underline{J} =6.5), 3.46 (dd, 1H, \underline{J} =10.8, 8), 3.02 (m, 2H), 1.20 (t, 3H, \underline{J} =6.5).
9. Kido, F.; Maruta, R.; Tsutsumi, K.; Yoshikoshi, A. Chem. Lett., **1979**, 163. Kido, F.; Tsutsumi, K.; Maruta, R.; Yoshikoshi, A. J. Am. Chem. Soc., **1979**, 101, 6420. Kido, F.; Noda, Y.; Maruyama, T.; Kabuto, C.; Yoshikoshi, A. J. Org. Chem., **1981**, 46, 4264. Kido, F.; Noda, Y.; Yoshikoshi, A. J. Chem. Soc., Chem. Commun., **1982**, 1209. Kido, F.; Tooyama, Y.; Noda, Y.; Yoshikoshi, A. Chem. Lett., **1983**, 881. Kido, F.; Noda, Y.; Yoshikoshi, A. Tetrahedron, **1987**, 43, 5467.

(Received in Japan 21 January 1989)